RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College under University of Calcutta)

B.A./B.SC. FIFTH SEMESTER EXAMINATION, DECEMBER 2012

THIRD YEAR INDUSTRIAL CHEMISTRY (Honours)

Date : 19/12/2012 Time : 11 am - 1 pm

Paper : V

Full Marks : 50

[4×5]

[1]

[Use separate Answer Book for each group]

Group - C

1. Answer **any four**:

- a) What are the major air pollutants in cites and what are the major sources of these pollutants?
- b) Describe with a flow sheet, air pollution control in cast iron industry.
- c) What steps have been taken to minimize automobile pollution?
- d) i) What is the name and chemical structure of the gas responsible for Bhopal disaster?
 - ii) What are the major green house gases? What does the term green house gas mean?
- e) Describe steps of industrial waste water treatment with a flow diagram.
- f) Explain why the bag filter is not suitable for particles form
 - i) lime slaking
 - ii) fly ash from coal burning
 - iii) heavy mist from chemical industry.

<u>Group – D</u>

<u>Unit – I</u>

(Answer <u>any four</u> questions)

2.	a)	Distinguish between dihedral angle and torsion angle with example.	[1]
	b)	Draw the potential energy diagram of n-butane due to rotation about $C_2 - C_3$ bond.	[2]
	c)	Between cis and trans 4-tertiary butyl cyclohexyl tosylates, which one would solvolyse at a faster rate and why?	[2]
3.	a)	Write down the products with stereochemistry, of the following reactions :	[3]
		i) (R)-MeCHBrCH ₂ CH ₃ \longrightarrow	
		ii) (R) – PhCHMeCl $\xrightarrow{H_2O}$	
	b)	'1,2-dihalocyclohexane always prefers to adopt diaxial conformation rather than diequatorial conformation' — give an explanation.	[2]
4.	a)	Draw the Newmann projection formulas of axial and equatorial methyl cyclohexanes and indicate the gauche-butane interactions in each formula due to methyl group.	[3]
	b)	'Gauche is the most stable conformer of 1,2-dihydroxy ethane' —explain.	[2]
5.	a)	What is ring inversion of cyclohexane? What are the possible pathways of inversion? Show any one path way of such inversion.	[3]
	b)	Show all possible conformers of cis-1, 3-diethyl cyclohexane. Indicate, which one is the most stable form and why?	[2]
6.	a)	Write down the product(s) with stereochemistry of the following reaction :	[2]
		$(R,R) - 2,3$ -dibromobutane $\xrightarrow{alcoholic}_{KOH, \Delta}$?	
	b)	'Trans-4-tertiary butyl cyclohexyl tosylate does not undergo base catalysed E2-reaction but the cis- isomer does' —Explain	[2]

c) Define the term 'conformation'.

7. Write down the chair conformations of each of cis and trans-1, 2-dimethyl cyclohexane and discuss their relative stability and indicate their relationships. Between cis and trans isomers, which one is more stable?
[2+2+1]

<u>Unit – II</u>

(Answer any two questions)

- 8. a) Separate the components from a mixture of ethyl amine and diethyl amine, showing the reactions involved. [2]
 - b) Using Gabriel phthalimide method, synthesise ethyl amine.
- 9. a) Synthesise 2-Ethyl naphthalene from benzene by the Haworth method. (Use any other organic reagent you need.) [2]
 - b) What happens when methyl amine is warmed with carbon disulfide and the resulting product is subsequently treated with mercuric chloride? What is the name of the entire reaction? $[1\frac{1}{2}+\frac{1}{2}]$

[3]

[2]

[3]

[2]

[3]

- c) What happens when naphthalene is heated with a mixture of conc. HNO_3 and H_2SO_4 acid? [1]
- 10. a) What happens when accetaldehyde is treated with ammonium formate? What is the name of this reaction?
 - b) Carry out the following conversions :
 - i) Benzene \rightarrow m-nitroaniline
 - ii) Aniline \rightarrow p-dinitrobenzene
- 11. a) Identify the compounds 'A' through 'D' in the following sequence of reactions :

$$A \xrightarrow{\text{conc. HNO}_3}_{\text{conc. H}_2\text{SO}_4} \longrightarrow \bigoplus_{Me}^{\text{NO}_2} \xrightarrow{\text{Sn/HCl}} B \xrightarrow{\text{NaNO}_2 + \text{HCl}}_{0.5^{\circ}\text{C}} C \xrightarrow{\text{D, heat}} \bigoplus_{Me}^{\text{F}}$$

b) Complete the following reactions :

i)
$$\underset{O_2N}{\overset{NH_2}{\longrightarrow}} \xrightarrow{NaNO_2+HCl} ? \xrightarrow{CuCl, HCl} ?$$

ii)
$$(\bigcirc NH_2 \\ NaNO_2 + HCl \\ 0.5^{\circ}C ? (KI, heat) ?$$

Me $(\bigcirc O) \\ Me \\ N_2Cl \\ \beta-naphthol ? (KI, heat) ?$

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